	me	Period	Date
	Home Work	401: Intro to Pe	riodic Table
1	In his periodic table, Mendeleev did not always list elements in order of increasi atomic mass because he wanted to group elements with similar		
a.	Properties	c.	Densities
b.	Atomic numbers	d.	Colors
2	Mendeleev arranged the	ne elements by increasing	atomic:
a.	Number	c.	Color
b.	Mass	d.	Density
3	A few elements were out of place on Mendeleev's periodic table. Name the scie who put them in the right place		
	Mendeleev	c.	Moseley
	Democritus	d.	Dalton
	when arranged in orde	-	roperties occurred at regular inte
	Atomic mass		
c.	Density		
	Color		
	Atomic number		
	behave similarly?	ements in a column of the	e periodic table that makes them
a.	behave similarly? Atomic number	ements in a column of the	e periodic table that makes them
a. b.	behave similarly? Atomic number Atomic mass	ements in a column of the	e periodic table that makes them
a. b. c.	behave similarly? Atomic number Atomic mass Electron distribution	ements in a column of the	e periodic table that makes them
a. b. c.	behave similarly? Atomic number Atomic mass	ements in a column of the	e periodic table that makes them
a. b. c. d.	behave similarly? Atomic number Atomic mass Electron distribution Radioactivity		e periodic table that makes them
a. b. c. d.	behave similarly? Atomic number Atomic mass Electron distribution Radioactivity The discovery of the n	oble gases changed Men	
a. b. c. d. 6a.	behave similarly? Atomic number Atomic mass Electron distribution Radioactivity The discovery of the n	oble gases changed Menc.	deleev's periodic table by adding
a. b. c. d. 6a.	behave similarly? Atomic number Atomic mass Electron distribution Radioactivity The discovery of the n Period	oble gases changed Menc c. d.	deleev's periodic table by adding Group
a. b. c. d. 6. a. b.	behave similarly? Atomic number Atomic mass Electron distribution Radioactivity The discovery of the n Period Series	oble gases changed Mence. c. d. law states that:	deleev's periodic table by adding Group Sublevel
a. b. c. d. 6. a. b.	behave similarly? Atomic number Atomic mass Electron distribution Radioactivity The discovery of the n Period Series Then modern periodic No two electrons can have the	oble gases changed Mence. c. d. law states that: ne same spin in the same	deleev's periodic table by adding Group Sublevel atomic orbital
a. b. c. d. 6a. b. 7a. b.	behave similarly? Atomic number Atomic mass Electron distribution Radioactivity The discovery of the n Period Series Then modern periodic No two electrons can have the series the physical and chemical p	oble gases changed Menoc. c. d. law states that: ne same spin in the same roperties of the elements	deleev's periodic table by adding Group Sublevel atomic orbital are functions of their atomic num
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